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Various State Education Boards. (A) Basic Laboratory Techniques - 1. To cut a glass tube or glass rod, 2. To bend the glass rod at an angle, 3. To draw a glass jet from a glass tube, 4. To bore a cork and fit a glass tube into it. (B) Characterisation and Purification of Chemical Substances- 1. To determine the melting point of the given unknown organic compound and its identification (simple laboratory technique), 2. To determine the boiling point of a given liquid when available in small quantity (simple laboratory method), 3. To prepare crystals of pure potash alum [K2SO4.Al2(SO4)3.24H2O]

from the given impure sample, 4. To prepare the pure crystals of copper sulphate from the given crude sample, 5. To prepare pure crystals of benzoic acid from a given impure sample. (C) Measurement of pH Values 1. To determine the pH value of vegetable juices, fruit juices, tap water and washing soda by using universal pH paper, 2. To determine and compare the pH values of solutions of strong acid (HCI) and weak acid (CH3COOH) of same concentration, 3. To study the pH change in the titration of strong base Vs. strong acid by using universal indicator paper, 4. To study the pH change by common ion (CH3COO- ion) in

case of weak acid (CH3COOH). 5. To determine the change in pH value of weak base (NH4OH) in presence of a common ion (NH4+), (D)Chemical Equilibrium 1. To study the shift in equilibrium between ferric ions and thiocvanate ions by changing the concentrations of either of the ions, 2. To study the shift in equilibrium between [Co(H2O)6]2+ and Cl- ions by changing the concentrations of either of the ions, (E) Quantitative Analysis 1. To prepare M/10 oxalic acid solution by direct weighing method, 2.To prepare M/10 solution of sodium carbonate by direct weighing method, 3.To determine the strength of

given solution of sodium hydroxide by titrating it against N/10 or M/20 solution of oxalic acid. 4.To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N/10 or M/20 sodium carbonate solution, (F) Qualitative Analysis 1. Analysis of Anions, 2. Analysis of Cations (G) Detection of Elements in Organic Compounds 1.To detect the presence of nitrogen, sulphur and halogens in a given organic compound by Lassaigne's test, 2. To detect the presence of nitrogen, sulphur and halogens in the given organic compound sample number by Lassaigne's test

INVESTIGATORY PROJECTS (A) Checking of Bacterial Contamination in Water 1.To. check the bacterial contamination in drinking water by testing sulphide ions (B) Methods of Water Purification 1.To purify water from suspended impurities by using sedimentation, 2. To purify water by boiling, 3.To purify water by distillation method, 4.To purify water by reverse osmosis technique. 5.To purify water by GAC method, 6.To purify water by bleach treatment, 7.To purify water by oxidising agent, 8.To purify water by ozone treatment method. (C) Water Analysis 1. To test the hardness of different water samples. (D)

Foaming Capacity of Various Soaps 1.To compare the foaming capacity of different washing soaps, 2.To study the effect of addition of sodium carbonate on foaming capacity of washing soap (E) Tea Analysis 1. To study the acidity of different samples of tea leaves (tea) by using pH paper (F) Analysis of Fruits and Vegetable Juices 1. To analyse the fruit and vegetable juices for the constituent present in them (G) Rate of Evaporation 1. To study the rate of evaporation of different liquids (H) Effect of Acids and Bases on Tensile Strength of Fibres 1.To compare the tensile strength of natural fibres and synthetic fibres, 2.To study the

effect of acids and bases on tensile strength of different fibres. Log & Antilog Table Mathematics Lab Manual Class X | According to the latest CBSE syllabus and other State Boards following the CBSE curriculum Nov 16 2022 With the NEP 2020 and expansion of research and knowledge has changed the face of education to a great extent. In the Modern times. education is not just constricted top the lecture method but also includes a practical knowledge of certain subjects. This way of education helps a student to grasp the basic concepts and principles. Thus, trying to break the stereotype that subjects like

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- <u>Lab Manual Science</u>
 Class 10
- <u>Lab Manual Social</u>
 <u>Science Class 10</u>
- Science Lab Manual
 Class X Follows The
 Latest CBSE Syllabus
 And Other State Board
 Following The CBSE
 Curriculam
- Mathematics Lab Manual

- Class X According To The Latest CBSE Syllabus And Other State Boards Following The CBSE Curriculum
- Core Science Lab Manual With Practical Skills For Class X
- Oswaal CBSE One For All Social Science Class 10 Reduced Syllabus For 2021 Exam
- Oswaal CBSE One For All Mathematics Standard Class 10 Reduced Syllabus For 2021 Exam
- Oswaal CBSE Question
 Bank Social Science
 Class 10 Reduced
 Syllabus For 2021 Exam
- <u>Mathematics Lab Manual</u> <u>Class IX According To</u>

The Latest CBSE Syllabus And Other State Boards Following The CBSE Curriculum

Class IX As Per The
Latest CBSE Syllabus
And Other State Board
Following The
Curriculum Of CBSE

• Science Lab Manual

- Oswaal CBSE One For All Science Class 10 For 2023 Exam
- Oswaal CBSE One For All Class 10 English Science Social Science Math Basic Set Of 4 Books For 2023 Exam
- Oswaal CBSE One For All Class 10 English Science Social Science Math Standard Set Of 4 Books

For 2023 Exam

- Oswaal CBSE MCQs
 Chapterwise For Term I
 II Class 10 Set Of 4
 Books Mathematics Basic
 Science Social Science
 - Sanskrit With The Largest MCQ Question
- Pool For 2021 22 Exam

 Oswaal CBSE MCOs
 - Chapterwise For Term I

 II Class 10 Set Of 4

 Books Mathematics Basic
 Science Social Science
 - With The Largest MCQ Question Pool For 2021

Computer Application

• Oswaal CBSE Question

22 Exam

Bank Mathematics Basic Class 10 Reduced Syllabus For 2021 Exam

- Practical Laboratory
 Manual Chemistry Class
 XI Based On NCERT
 Guidelines By Dr S C
- Rastogi Er Meera Goyal

 Index Of Supply Manuals
 Transportation Corps
- <u>Technical Manual</u>
- Oswaal CBSE Question
 Bank Class 9 English
 Language Literature
 Reduced Syllabus For
 2021 Exam
- Air Force Manual
- Oswaal NCERT CBSE
 Pullout Worksheet Class
 10 English Language
 Literature For 2021 22
 - <u>Exam</u>
- Comprehensive Lab Manual Science VII
- Teachers Manual Of

Method And Organisation Second Edition Re arranged Etc

- <u>Bureau Of Navigation</u>
 Manual
- Manual Of Classification
 Of Subjects Of Invention
 Of The United States
 Patent Office
- <u>GPO Classification</u> Manual
- Bairn CBSE Solutions Of RS Aggarwal
 Mathematics Class 10

For 2021 Exam

- Oswaal CBSE Question
 Bank Class 10 Science

 Oswaal E
 - For 2021 Exam
- Operators Manual
- <u>United States</u>
 <u>Department Of</u>
 Commerce Publications
- <u>Science Lab Manual</u>
- Manual NGB
- The United States
 Department Of
 Commerce Publications

Catalog And Index Supplement

- National Guard Bureau Manual
- Ordnance Safety Manual
- Statistics Of The
 Communications Industry
 In The United States
- <u>Customs Issuance System</u> Index
- <u>Catalog Of Copyright</u> <u>Entries Third Series</u>
- <u>United States Army</u> Training Manual